A nonlinear, control-oriented model for ionic polymer–metal composite actuators

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Abstract
Ionic polymer–metal composites (IPMCs) form an important category of electroactive polymers and have many potential applications in biomedical, robotic and micro/nanomanipulation systems. In this paper, a nonlinear, control-oriented model is proposed for IPMC actuators. A key component in the proposed model is the nonlinear capacitance of the IPMC. A nonlinear partial differential equation (PDE), which can capture the fundamental physics in the IPMC, is fully considered in the derivation of nonlinear capacitance. A systems perspective is taken to get the nonlinear mapping from the voltage to the induced charge by analytically solving the nonlinear PDE at the steady state when a step voltage is applied. The nonlinear capacitance is incorporated into a circuit model, which includes additionally the pseudocapacitance due to the electrochemical adsorption process, the ion diffusion resistance, and the nonlinear DC resistance of the polymer, to capture electrical dynamics of the IPMC. With electromechanical coupling, the curvature output is derived based on the circuit model. The proposed model is formulated in the state space, which will be the starting point for nonlinear controller design. Experimental verification shows that the proposed model can capture the major nonlinearities in the electrical response of the IPMC.

(Some figures in this article are in colour only in the electronic version)

1. Introduction

Electroactive polymers (EAPs) have gained their nickname artificial muscles because of their similarity to biological tissues and their built-in actuation capabilities [1]. Ionic polymer–metal composites are one category of ionic EAPs, where the actuation effect arises from electrically induced ion transport. An IPMC sample is made of a thin ion-exchange membrane (e.g. Nafion) sandwiched by two noble metal electrodes, such as platinum or gold [2]. When a small voltage is applied to an IPMC, the movement of hydrated ions leads to redistribution of charge densities inside the IPMC. The resulting electrostatic forces and the accompanying differential swelling are believed to lead to bending [1, 3, 4]. Because IPMCs are soft, resilient, biocompatible, and capable of producing large deformation under a low action voltage, they have many potential applications in biomedical devices [5], biological studies [6], and biomimetic robots [7, 8].

An accurate and practical mathematical model is desirable in the application of IPMCs. Extensive work has been done in modeling of IPMCs. Current modeling work can be classified into three categories based on their complexity levels. Based purely on the empirical responses, black-box models, e.g. [9], offer minimal insight into the governing mechanisms within the IPMC. As a more detailed approach, the gray-box models, e.g. [10], are partly based on physical principles while also relying on empirical results to define some of the more complex physical processes. In the most complex form, white-box models with partial differential equations (PDEs), e.g. [3], attempt to explain the underlying physics for the sensing and actuation responses of IPMCs, but they are not practical for real-time control purposes. Del Bufalo et al applied mixture theory to the modeling of IPMC actuators [11], but they only addressed the behavior under quasi-static actuation. Chen and co-workers [12, 13] bridged the gap between the empirical models and the physical models for IPMC sensors and actuators, by exactly solving the governing PDEs in the Laplace domain, and coming up with an infinite-dimensional
transfer function relating the output to the input. However, this control-oriented model was based on a linear PDE which ignored the nonlinear terms in the physical model. This approximation only holds when a small voltage is applied. A nonlinear model is needed when a relatively high voltage is applied to the IPMC to generate large deformation.

Nonlinear behaviors of IPMCs have been reported in the literature. Chen et al [14] employed the Preisach operator to capture the hysteresis in IPMCs. Bonomo et al [15] reported a nonlinear circuit model of IPMCs. However, it is an empirical nonlinear model and does not capture the fundamental physics in IPMCs. Nemat-Nasser [4] captured a nonlinear capacitance of IPMCs with an assumption that there is asymmetric charge distribution along the thickness direction and that a cation-depleted layer forms near the anode side when a relatively high voltage is applied. However, the region without cation depletion was still governed by the linear PDE. Porfiri [16] applied Poisson–Nernst–Planck equations to investigate charge dynamics in IPMCs. He used matched asymptotic expansions to solve the nonlinear PDE and obtained a nonlinear circuit model for IPMCs, which includes a nonlinear capacitance and a linear diffusion resistance. However, the circuit model does not accommodate electrochemical reaction near the electrodes and DC resistance of the polymer. No actuation model was further derived based on the circuit model.

Our proposed model is based on the original nonlinear PDE, which can capture ion diffusion, ion migration, and electrostatic interactions in IPMCs. The modeling work starts from the analysis of the equilibrium of IPMCs under a step voltage input, which can be captured by a nonlinear ordinary differential equation (ODE). Numerical analysis of the nonlinear ODE demonstrates an asymmetric charge distribution along the thickness direction. The nonlinear term in the original PDE cannot be ignored when a moderate voltage (>0.2 V) is applied. Since the nonlinear ODE cannot be explicitly solved, a systems perspective is taken to derive the analytical nonlinear mapping from the voltage to the charge. It is verified by the numerical solution, and is practically useful in real-time control.

A nonlinear circuit model is employed to capture the electrical dynamics of IPMCs. It incorporates nonlinear capacitance of IPMCs derived from the nonlinear mapping function between the charge and the voltage, ion diffusion resistance [15, 16], pseudocapacitance due to the electrochemical process at the polymer–metal interface [17], and nonlinear DC resistance of the polymer [15]. Based on the electromechanical coupling effect, the curvature output can be obtained from the electrical dynamic model. The proposed model shows consistency with the linear model when the voltage is small [3]. With definitions of the state variable, input, and output, the model is further presented in the state space, which will be the starting point for nonlinear control design. Parameters are measured or identified through experiments. The proposed model is validated in experiments.

The remainder of the paper is organized as follows. The governing nonlinear PDE is reviewed in section 2. Section 3 shows numerical and analytical procedures for understanding the PDE at the steady state. In section 4, a nonlinear circuit model is introduced to capture the electrical dynamics of IPMCs. The derivation of curvature output of the IPMC and a nonlinear control-oriented model are also shown in section 4. Experimental validation of the proposed model is presented in section 5. Finally, concluding remarks are provided in section 6.

A preliminary version of the paper was presented at the 2008 IEEE Conference on Decision and Control [18].

2. The governing nonlinear PDE

The governing PDE for charge distribution in an IPMC was first presented in [3] and then used by Farinholt [19] for investigating the actuation and sensing responses. Let \( D, E, \phi \), and \( \rho \) denote the electric displacement, electric field, electric potential, and charge density, respectively. The following equations hold:

\[
E = \frac{D}{\kappa_e} = -\nabla \phi, \quad (1)
\]

\[
\nabla \cdot D = \rho = F(C^+ - C^-), \quad (2)
\]

where \( \kappa_e \) is the effective dielectric constant of the polymer, \( F \) is Faraday’s constant, and \( C^+ \) and \( C^- \) are the cation and anion concentrations, respectively. Since the thickness of an IPMC is much smaller than its length or width, we can assume that \( D \) and \( E \) are restricted to the thickness direction ( \( x \)-direction) only, and from here on drop the boldface notation for these variables. Figure 1 shows the geometric definitions of the IPMC.

The continuity expression relates the ion flux \( J \) (in the \( x \)-direction) to the cation concentration \( C^+ \):

\[
\frac{\partial J}{\partial x} = -\frac{\partial C^+}{\partial t}. \quad (3)
\]

The ion flux consists of diffusion, migration, and convection terms, and can be expressed as [3, 13]

\[
J = -\frac{d \kappa_e}{F} \left( \frac{\partial^2 E}{\partial x^2} - \frac{F(1 - C^- \Delta V)}{RT} E \left( \frac{\partial E}{\partial x} + \frac{F C^-}{\kappa_e} \right) \right), \quad (4)
\]

where \( d \) is the ionic diffusivity, \( R \) is the gas constant, \( T \) is the absolute temperature, and \( \Delta V \) is the volumetric change. With the continuity equation (3), one can derive the nonlinear governing PDE in terms of the electric field,

\[
\frac{\partial^2 E}{\partial t \partial x} = d \left( \frac{\partial^3 E}{\partial x^3} - \frac{F(1 - C^- \Delta V)}{RT} \left[ \frac{\partial^2 E}{\partial x^2} \right]^2 \frac{\partial E}{\partial x} \right) - \frac{F^2 C^- (1 - C^- \Delta V) \partial E}{RT \kappa_e} \left( \frac{\partial E}{\partial x} \right)^2. \quad (5)
\]

![Figure 1. Geometric definitions of an IPMC beam.](Image 514x514 to 535x535)
In several papers [3, 4, 12, 19], the nonlinear term involving \(\frac{dE}{dx} \cdot E\) in (4) has been ignored based on the assumption
\[
\rho(x) = \kappa e \frac{\partial E}{\partial x} \ll C^{-} F,
\]
resulting in a linear PDE,
\[
\frac{d\rho}{dt} - d \frac{d^2 \rho}{dx^2} + \frac{F^2 dC^{-}}{\kappa_e RT} (1 - C^{-} \Delta V) \rho = 0.
\]
However, the assumption (6) will not hold when a relatively high voltage is applied, as to be shown later in this paper.

### 3. Analysis of the PDE at the steady state

A key component in the proposed model is the capacitance of the IPMC. The first step in our modeling work is to analyze the nonlinear PDE (5) at equilibrium when a step voltage is applied, and find the mapping function from voltage to charge. At equilibrium, \(J = 0\). Equation (4) then implies
\[
E''(x) = \frac{\rho(x)}{\kappa_e RT} E'(x) - \frac{F^2 C^{-}}{\kappa_e RT} E(x) = 0,
\]
which becomes
\[
E''(x) - a E'(x) E(x) - b E(x) = 0.
\]
The following two conditions hold for the ODE.

1. The overall charge-balance condition leads to
\[
\int_{-h}^{h} \rho(x) \, dx = 0.
\]

2. The potential difference is equal to the applied voltage
\[
\int_{-h}^{h} E(x) \, dx = V.
\]

\[\text{Table 1. Parameters in the model.}\]

<table>
<thead>
<tr>
<th>(F)</th>
<th>(R)</th>
<th>(T)</th>
<th>(R_e)</th>
</tr>
</thead>
<tbody>
<tr>
<td>96.487 C mol(^{-1})</td>
<td>8.3143 J mol(^{-1}) K(^{-1})</td>
<td>300 K</td>
<td>18 (\Omega)</td>
</tr>
<tr>
<td>48 (\Omega)</td>
<td>22 mm</td>
<td>10 mm</td>
<td>100 (\mu)m</td>
</tr>
<tr>
<td>1091 mol m(^{-3})</td>
<td>1.34 \times 10(^{-6}) F m(^{-1})</td>
<td>1 \times 10(^{-6}) mol m(^{-3})</td>
<td>4 \times 10(^{3})</td>
</tr>
<tr>
<td>0.129 J C(^{-1})</td>
<td>0.56 GPa</td>
<td>210 (\mu)C cm(^{-2})</td>
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</tr>
<tr>
<td>(a_0)</td>
<td>(Y_1)</td>
<td>(q_1)</td>
<td></td>
</tr>
<tr>
<td>(1 \times 10^{-5}) A V(^{-1})</td>
<td>(1 \times 10^{-4}) A V(^{-2})</td>
<td>(1 \times 10^{-4}) A V(^{-3})</td>
<td></td>
</tr>
</tbody>
</table>

### 3.1. Numerical solution of the PDE at the steady state

In order to numerically solve the second-order ODE (8), one needs to know the boundary conditions \(E(-h)\) and \(E'(h)\), or \(E(h)\) and \(E'(h)\). However, these conditions are unknown. So we change the initial point to \(x = x_0\), where \(x_0\) is defined as the zero charge density point, \(\rho(x_0) = \kappa_e E'(x_0) = 0\). Note that \(x_0\) will also depend on the applied voltage. In order to satisfy the conditions (11) and (12), we run the following steps recursively.

**Step 1:** Assign a value to \(x_0\) such that \(-h < x_0 < +h\), and a very small value to \(E(x_0) = E_0\).

**Step 2:** Integrate ODE (8) forward from \(x = x_0\) to \(h\) and backward from \(x = x_0\) to \(-h\), separately, to get \(E(x)\) and \(E'(x)\).

**Step 3:** If \(|E(h) - E(-h)| \leq \epsilon\), go to step 4; if \(E(h) > \epsilon\), decrease \(x_0 = x_0 - \epsilon_2\) and go to step 2; if \(E(h) < -\epsilon\), increase \(x_0 = x_0 + \epsilon_1\) and go to step 2.

**Step 4:** Do integration \(-\int_{x_0}^{h} E(x) \, dx\) to get \(\phi(x)\) with \(\phi(-h) = 0\).

**Step 5:** If \(|\phi(h) - V| < \epsilon_2\), then go to step 6; if \(\phi(h) - V > \epsilon_2\), then increase \(E_0 = E_0 + \epsilon_1\) and go to step 1; if \(\phi(h) > V < \epsilon_2\), then decrease \(E_0 = E_0 - \epsilon_3\) and go to step 1.

**Step 6:** Calculate \(\rho(x) = \kappa_e E'(x)\) and do integration \(Q = \int_{x_0}^{h} \rho(x) S \, dx\), then stop.

In the steps above, \(\epsilon, \epsilon_1, \epsilon_2, \text{ and } \epsilon_2\) are small positive constants; \(S = WL\) is the surface area of the IPMC. All the physical parameters in the PDE are listed in table 1 and the identification of these parameters is described in section 5. Figure 2 shows the numerical simulation results when \(V = 2.61\) V. Figure 2(a) shows the asymmetric charge distribution along the thickness direction. Two inset figures show the details at the turning points of the curve. The negative charge density near the anode approaches the saturation value \(C^{-} F\). The charge density distribution also shows that (6) will not hold in the region close to the boundaries. In other words, the linear PDE will not hold when a high voltage is applied.

Numerical solution offers us insight into the charge distribution, electrical field and electrical potential along the
It is also the starting point in the derivation of nonlinear capacitance of IPMC.

Similarly, if we take the systems perspective to solve this problem, what we are really concerned about is how the total charge is analytically related to the input voltage. Then it is not necessary to know the explicit function $E(x)$. By integrating both sides of (13) from $x = -h$ to $h$, we get

$$\int_{-h}^{h} E''(x) \, dx - \int_{-h}^{h} aE'(x) \, dx - \int_{-h}^{h} bE(x) \, dx = 0.$$  \hfill (16)

With (12), equation (16) can be written as

$$E'(h) - E'(-h) - \frac{a}{2} (E'(h) - E'(h)) - bV = 0.$$  \hfill (17)

From (11) and (17), one can get

$$V = \frac{1}{b} (E'(h) - E'(-h)).$$  \hfill (18)

The total charge $Q$ can be obtained by integrating $\rho(x)$ from $x = x_0$ to $h$, where $\rho(x_0) = 0$:

$$\int_{x_0}^{h} \rho(x) \, dx = Q = (E(h) - E(x_0)) S K_e.$$  \hfill (19)

In section 3.1, figure 2(b) shows that $E(x_0) \ll E(h)$ ($E(x_0) = 1.5 \times 10^{-23}$ V m$^{-1}$ and $E(h) = 2 \times 10^7$ V m$^{-1}$). So (19) can be written as

$$E(h) = E(-h) = \frac{Q}{S K_e}.$$  \hfill (20)

Note that $E'(h)$ and $E'(-h)$ are the positive root and negative root of (14), respectively, when $y = E(h)$.

Define

$$k \triangleq \frac{1}{2} \left( \frac{Q}{S K_e} \right)^2.$$  \hfill (21)

Then (14) can be written as

$$\frac{p}{a} - k = \frac{b}{a^2} \ln \left( \frac{a}{b} p + 1 \right).$$  \hfill (22)
Figure 3. Illustration of solving for $p_1$ and $p_2$.

**Proposition 3.1.** If $k > 0$, there exist two roots $(p_1, p_2)$ for (22) such that $-\frac{b}{a} < p_1 < 0$ and $p_2 > 0$. Furthermore, if $k = 0$, then $p_1 = p_2 = 0$.

See appendix A for the proof of Proposition 3.1.

In order to get the mapping function from $V$ to $Q$, we need find out how $k$ is related to the distance of two roots $|p_1 - p_2|$. As shown in figure 3, the roots of (22) are the intersection points $(p_1$ and $p_2$) of the following two curves:

$$\eta = \frac{b}{a^2} \ln \left( \frac{a}{b} p + 1 \right) = f(p),$$

$$\eta = \frac{p}{a} - k = \lambda(p).$$

From figure 3, the negative root $p_1$ will never hit the line $p = -\frac{b}{a}$ because $p = -\frac{b}{a}$ is the asymptote of the logarithmic function of $\eta = \ln(\frac{a}{b} p + 1)$. The physical explanation of this is the following. Since the negative ions cannot move and the negative ion density is uniform in the IPMC, $\frac{\kappa e}{C - F}$ is the bound of the negative charge density. So $p > -\frac{b}{a}$ implies that a layer of depleted positive charges will not form, although the positive charge density can be very close to zero.

Finally, we can write $k$ in terms of the voltage input $V$

$$k = \Gamma(V) \triangleq \begin{cases} \Gamma(V), & V > 0 \\ 0, & V = 0 \end{cases} \quad (23)$$

where

$$\Gamma(V) \triangleq \frac{b}{a^2} \left( \frac{aV}{e^{aV} - 1} - \ln \left( \frac{aV}{e^{aV} - 1} \right) - 1 \right). \quad (24)$$

and $\Gamma$ is continuous at $V = 0$. See appendix B for the derivation of (23). With (21) and (23), one can get the total charge $Q$ as a function of $V$:

$$Q = S\kappa_{e} \sqrt{2\Gamma(V)}. \quad (25)$$

Note that the nonlinear mapping function from the input voltage is similar to the one reported in [16], which Porfiri obtained by solving the nonlinear PDE with matched asymptotic expansions.

When $V \to 0$, one can approximate $\Gamma(V)$ using its Taylor series expansion around $V = 0$,

$$\Gamma(V) \approx \frac{b|V|^2}{8}. \quad (26)$$

Then $Q \approx \frac{\sqrt{2} b S \kappa_{e}}{8}$, which is consistent with the charge generated in the linear case [3].

Figure 4(a) shows the simulation results of charge versus voltage at the steady state. It includes the results based on the analytical solution of the nonlinear ODE, the numerical solution of the nonlinear ODE, and the numerical solution of the linear ODE (which ignores the nonlinear term), respectively. One can see that the analytical solution matches well the numerical solution of the nonlinear ODE. When the voltage is small, one can ignore the nonlinear term in the nonlinear PDE. However, if a relatively large voltage ($>0.2$ V) is applied, the error between the nonlinear model and the linear one becomes significant. Figure 4(b) shows the charge densities at the boundaries of the IPMC based on the analytical solution. Note that these densities correspond to the solutions.
$p_1$ and $p_2$ of (22), which can be analytically obtained as shown in appendix B. The figure shows that $\rho(-h)$ cannot go lower than $-C^- F$, but $\rho(h)$ can go arbitrarily high. The inset figure shows the zoom-in view around the turning corner where the negative charge density approaches saturation. It is consistent with the numerical simulation results in figure 2(a). The voltage corresponding to the turning corner is about 0.2 V. The above analytical analysis of PDE at the steady state will be helpful for deriving the nonlinear capacitance of the IPMC, which will be discussed next.

4. Nonlinear circuit model

The above analysis can only capture the nonlinear capacitance of the IPMC. A dynamic model needs to capture the transient processes in the IPMC as well. We propose a nonlinear circuit model, as shown in figure 5. It incorporates the nonlinear capacitance of the IPMC $C_1$, pseudocapacitance $C_a$ due to the electrochemical adsorption process at the polymer–metal interface, ion diffusion resistance $R_c$, electrode resistance $R_a$, and nonlinear DC resistance of the polymer $R_{dc}$.

4.1. Nonlinear capacitance of the IPMC

The nonlinear capacitance can be obtained by taking the derivative of (25),

$$C_1(V) = \frac{dQ}{dV} = S \kappa \frac{\Gamma'(V)}{\sqrt{2 \Gamma(V)}},$$

where $\Gamma'(V)$ is the first derivative of $\Gamma(V)$. From (24),

$$\Gamma'(V) = \frac{b}{a} \left(1 - e^{aV} \right) \left( e^{aV} - 1 - aV e^{aV} \right)^2.$$

The proposed analytical solution of nonlinear capacitance captures the fundamental physics in the IPMC. It is represented by a function in terms of physical parameters and dimensions, and is geometrically scalable.

4.2. Pseudocapacitance due to adsorption

For an electrochemical surface process, e.g. the so-called underpotential deposition of H [20], the following holds:

$$M + H_3O^+ + e \rightleftharpoons MH_{ads} + H_2O,$$

where M is the substrate (usually a noble metal, Pt, Rh, Ru, or Ir). Since IPMC has Pt as electrode and some electrolyte in the polymer, the underpotential deposition process should be incorporated into the model [17]. The adsorption current due to this electrochemical process can be represented by [20]

$$I_a = C_a(V_a) \frac{dV_a}{dt},$$

where

$$C_a(V_a) \approx \frac{q_1 \sqrt{F}}{RT (K_{1c} e^{V} + e^{\frac{q_1}{a} V})^2}.$$ 

$V_a$ is the voltage on the pseudocapacitance, $q_1$ is some constant (for H on polycrystalline Pt, $q_1 = 210 \mu C cm^{-2}$ [20]), $K_1 = \frac{\kappa}{\kappa_1}$, $k_1$ and $k_{-1}$ are the chemical rate constants for forward and reverse directions of (29), and $e^{V}$ is the concentration of $H^+$.

4.3. Nonlinear DC resistance

The current response under a step voltage input will not vanish at the steady state [15] because of the DC resistance of the polymer. One can approximate the DC current by a series of polynomial functions $Y(V)$. In this modeling work, we use a third-order polynomial function:

$$I_{dc} = Y(V) \approx \text{sgn}(V)(Y_1|V| + Y_2|V|^2 + Y_3|V|^3).$$

Note that $I_{dc}$ is supposed to be an odd function of $V$. This is why $\text{sgn}(V)$ appears in (32).

4.4. Curvature output

The induced stress is proportional to the charge density [3]:

$$\sigma = \alpha_0 \rho,$$

where $\alpha_0$ is the coupling constant. The moment generated by IPMC can be written by

$$M = \int_{-h}^{h} \sigma x W dx = \int_{-h}^{h} W \alpha_0 x \rho(x) dx.$$ 

Since $\rho(x) = \kappa \frac{dE}{dx}$,

$$M = \int_{-h}^{h} W \alpha_0 \kappa x dE = \int_{-h}^{h} E d\left( W \alpha_0 \kappa x \right).$$

Since $E(-h) = E(h) = \frac{Q}{2 \kappa}$, with (25) and $\int_{-h}^{h} E dx = V$,

(35) can be written as

$$M = W \alpha_0 \kappa \left( \text{sgn}(V)2h \sqrt{2 \Gamma(|V|)} - V \right).$$

If one takes $V \to 0$,

$$M \to W \alpha_0 \kappa c V (h \sqrt{b} - 1) \approx W \alpha_0 \kappa c V h \sqrt{b},$$

which is consistent with the moment reported in the linear case [3]. One can then obtain the curvature output via

$$\kappa = \frac{M}{V^2 I_1}.$$
where \( I = \frac{2}{3} Wh^3 \) is the moment inertia and \( Y_e \) is the equivalent Young’s modulus of the IPMC. With (36), equation (38) can be written as
\[
\kappa = \Psi(V) = \frac{3\alpha_0 \kappa_e (\text{sgn}(V) 2h \sqrt{2Y_e ||V||} - V)}{2Y_e h^3}.
\] (39)

We note that one could also use nonlinear elasticity theory to model the mechanical output under large deformation [21]. This is outside of the scope of the current paper.

4.5. Nonlinear control-oriented model

The objective of this work is to derive a control-oriented nonlinear model which can be used in controller design. Based on the circuit model (figure 5) and the curvature output (36), the model structure is shown in figure 6.

From (30) and (32), one can get
\[
\frac{dV}{dt} = \frac{I - V}{R_a + R_c} - Y(V) \frac{dV}{C_1(V) + C_2(V)}.
\] (40)

Defining the state variable \( x = V \), the control input \( u = U \), and the system output \( y = \kappa \), one can obtain a first-order nonlinear dynamic model in the state space:
\[
\dot{x} = - \frac{x + Y(x)(R_a + R_c) - u}{(C_1(x) + C_2(x))(R_a + R_c)}, \quad y = \Psi(x).
\] (41)

5. Experimental verification

Some physical parameters can be directly measured through the experiments, such as the temperature \( T \), the dimensions \( L, W, h \), the equivalent Young’s modulus of the IPMC \( Y_e \), and the electrode resistance \( R_a \). Some parameters are physical constants, such as \( R, F, q_1 \). Other parameters \( (Y_1, Y_2, Y_3, R_c, K_1, C, \kappa_e, \alpha_0) \) can be identified by a model fitting process. To ignore the distributed resistance effect of the electrode on the impedance of the IPMC, we clamped the IPMC sample with two aluminum electrodes, which short circuit the electrode resistance \( R_a \). To measure the current, we added another small resistance \( R_i = \frac{18}{\Omega} \), which is serially connected with the IPMC. By measuring the voltage \( V_i \), one can obtain \( I_s = \frac{V_i}{R_i} \). Figure 7 illustrates the configuration of the experimental set-up. In the rest of this section, \( R_a \) is replaced by \( R_i \).

The current responses under a series of step voltage inputs are measured in order to identify these unmeasured parameters. Figure 8 shows the initial current \( I_{0} \) and the DC current \( I_{\infty} \) under step voltage inputs. The initial current \( I_{0} \) is the current...
when the capacitors are uncharged ($V = 0$). The DC current $I_{\infty}$ is the current when the capacitors have been fully charged. Since the polymer resistance $R_{dc}$ is much larger than $R_c$, when $V = 0$, one can get (see figure 5)

$$I_{\infty} \approx \frac{U}{R_i + R_c}.$$  \hfill (42)

Figure 8(a) shows that the initial current is a linear function of the step voltage. One can obtain the linear resistance $R_i + R_c$ by calculating the slope of $I_{\infty}$ versus $U$. Figure 8(b) shows that the DC current can be approximated by a third-order polynomial function of $V$ (32), thus the coefficients $Y_1$, $Y_2$, $Y_3$ can be identified. $e^{\bar{H}^+}$, $K_1$, $\kappa_c$ can be tuned to fit the transient process. Figure 9 shows the current responses under a step input of 1.0 V, suggesting that the proposed model can capture both the transient process and the steady state in current responses under a step voltage input. Identification of $\alpha_0$ was reported in [12]. Table 1 shows all the parameters in the model.

The model is further verified through an experiment to examine the current response under a sinusoid voltage input with frequency 0.01 Hz and amplitude 3 V. Figure 10 shows that the model can predict the current response well.

6. Conclusions and future work

In this work, a nonlinear, control-oriented model is proposed for IPMC actuation. It is derived based on the nonlinear dynamics governing the PDE. Numerical analysis of the nonlinear PDE at the steady state shows an asymmetric charge distribution along the thickness. A systems perspective is taken to obtain a nonlinear mapping from the voltage to the induced charge, which captures the nonlinear capacitance of the IPMC. A nonlinear circuit model is employed to represent the electrical dynamics of the IPMC, including the nonlinear capacitance of the IPMC, the ion diffusion resistance, the pseudocapacitance due to the electrochemical process at the polymer–metal interface, and the nonlinear DC resistance of the polymer. The proposed model is described in the state space, which will be the starting point for nonlinear control design. The proposed nonlinear electrical model is validated experimentally.

The effect of distributed surface resistance is ignored in the presented nonlinear circuit model. In practice, one can consider several segments of the circuit to capture the distributed surface resistance and achieve more precise modeling. However, it will lead to more state variables in the model and more computation in real-time control. This is a tradeoff between model accuracy and implementation complexity.

Future work will be focused on the following: (1) incorporating mechanical dynamics into the model; (2) design of nonlinear controller and nonlinear system analysis; (3) application to IPMC-actuated biomedical devices and biomimetic robots, which require large deformation of the IPMC.

Acknowledgments

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Appendix A. Proof of proposition 3.1

Proof. Define

$$\chi(p) = \frac{p}{a} - \frac{b}{a^2} \ln\left(\frac{a}{b} p + 1\right) - k, \quad p > -\frac{b}{a}. \hfill (A.1)$$

Let us start with the case when $k > 0$. Because $\chi(0) = -k < 0$, $\chi(p)$ is continuous in $\left(-\frac{b}{a}, +\infty\right)$ and

$$\lim_{p \to -\frac{b}{a}} \chi(p) = +\infty > 0, \quad \lim_{p \to +\infty} \chi(p) = +\infty > 0, \hfill (A.2)$$
there exist $p_1 \in (-\frac{b}{a}, 0)$, $p_2 \in (0, +\infty)$ such that $\chi(p_1) = 0$ and $\chi(p_2) = 0$. Since

$$\chi'(p) = \frac{p}{ap + b},$$  \hspace{1cm} (A.3)

with $p > -\frac{b}{a}$, we can get $\chi'(p) > 0$ when $p > 0$ and $\chi'(p) < 0$ when $0 > p > -\frac{b}{a}$. So $\chi(p)$ is monotonically increasing in $(0, +\infty)$, and monotonically decreasing in $(-\frac{b}{a}, 0)$. Then $p_1$ and $p_2$ should be unique. When $k = 0$, $\chi(0) = 0$, implying $p_1 = p_2 = 0$. \hfill \Box$

### Appendix B. Derivation of equation (23)

Define $D = p_2 - p_1$. Equation (18) can then be written as $V = \frac{D}{b}$. From figure 3,

$$\frac{p_2 - p_1}{\eta_2 - \eta_1} = a,$$  \hspace{1cm} (B.1)

where $\eta_1$ and $\eta_2$ are the $\eta$-coordinates corresponding to $p_1$ and $p_2$, respectively, in figure 3. Equation (B.1) implies

$$\frac{D}{\ln(\frac{p_2}{a} + 1) - \ln(\frac{p_1}{a} + 1)} = a.$$  \hspace{1cm} (B.2)

With $p_2 = D + p_1$ and (B.2), one can solve $p_1$ in terms of $D$,

$$p_1 = \begin{cases} \frac{D}{e^{x_D} - 1} - \frac{b}{a}, & D > 0 \\ 0, & D = 0. \end{cases}$$  \hspace{1cm} (B.3)

When $D \to 0$, with l’Hospital’s rule,

$$\lim_{D \to 0} \left( \frac{D}{e^{x_D} - 1} - \frac{b}{a} \right) = \lim_{D \to 0} \left( \frac{1}{e^{x_D}} \right) - \frac{b}{a} = 0.$$  \hspace{1cm} (B.4)

So $p_1$ is still a continuous function of $D$. Since

$$\frac{b}{a^2} \ln \left( \frac{a}{b} \left( \frac{D}{e^{x_D} - 1} - \frac{b}{a} \right) + 1 \right) = \frac{p_1}{a} - k,$$  \hspace{1cm} (B.5)

from (B.3), one can get

$$\frac{b}{a^2} \ln \left( \frac{a}{b} \left( \frac{D}{e^{x_D} - 1} - \frac{b}{a} \right) + 1 \right) = \frac{1}{a} \ln \left( \frac{D}{e^{x_D} - 1} - \frac{b}{a} \right) - k.$$  \hspace{1cm} (B.6)

Since $D = bV$, one can get equation (23) from equation (B.6). Note that with l’Hospital’s rule,

$$\lim_{V \to 0} \left( \frac{aV}{e^{xV} - 1} \right) = \lim_{V \to 0} \left( \frac{a}{ae^V} \right) = 1,$$  \hspace{1cm} (B.7)

one gets

$$\lim_{V \to 0} \left( \frac{b}{a^2} \left( \frac{aV}{e^{xV} - 1} - \ln \left( \frac{aV}{e^{xV} - 1} \right) - 1 \right) \right) = 0.$$  \hspace{1cm} (B.8)

So $k$ is a continuous function of $V$.

### References


